



Development and Characterization of Nanodiamond-Doxorubicin (DOX) Conjugates for Enhanced Delivery against Breast Cancer Cells

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ABSTRACT

In this work, we have introduced a carbon nanomaterial (nanodiamond), to bind with anticancer drug doxorubicin (DOX) with via amide bond conjugation for cancer drug delivery and therapy. Nanodiamonds (ND) was initially carboxylated by the surface modification along the treatment with strong alkaline solution (H₂SO₄:HNO₃) and then activated the carboxyl moiety of ND with the addition of EDC. Anticancer drugs were bound to the ND through a succession of chemical modifications by adipic acid dihydrazide (ADH). The ND-Drug conjugate was analyzed by Nuclear Magnetic Resonance (¹H-NMR) Spectroscopy, Fourier Transform Infrared (FTIR) Spectroscopy and Mass Spectroscopy (MS), Atomic Force Microscopy (AFM), Particle size, Zeta potential, Drug release, SRB assay against MCF-7 cells, and DNA fragmentation. Spectroscopic analysis confirms the conjugation of nanodiamond with different anticancer drug. AFM photomicrograph represents the surface morphological features of ND-DOX conjugates. *In-vitro* investigation showed that ND-DOX conjugates have slow and sustained drug release characteristics. *In-vitro* cytotoxicity studies, an enormous cytotoxic potential of ND- Drug conjugates were showed against cancer cell line. Above all findings were suggested that the ND-DOX conjugates may be a potential inhibitor of MCF-7 cancer cells to act as a drug candidate. According to all these data it can be confirm that the ND-DOX conjugates could be an effective agent for drug delivery and could be promising in future for tumor targeting strategy.

INTRODUCTION

As a major cause of mortality, cancer remains a global public health concern. To date, the most common treatment of cancer has been chemotherapy, the therapeutic effect of which is far from optimal due largely to the nonspecific toxicity of chemotherapeutics. That is why the idea of cancer nanotechnology is put forward, which provides a unique approach against cancer by applying nanotechnology in cancer management (Jones, 2007; Saenz *et al.*, 2014). Cancer is one of the pre-eminent causes of disease and infection globally. Lung, colorectal, stomach, breast cancer etc. are the mostly diagnosed cancer worldwide reported by World Health Organisation (Jemal *et al.*, 2011). Beside these, currently the lung cancer is more common cancer cause and accounting about 16.7% diagnosed in men. In the past, the most common cause for lung cancers has been associated with smoking (U.S. Department of Health and Human Services, 2004). Chemotherapy is

the first-line therapy to diagnose and treat breast, liver & lung cancer at advanced stage, in which the drug is administered through intravenous route to reach the drug in systemic circulation for effective therapy (Paumier and Péchoux, 2010; Arriagada *et al.*, 2004). Nanodiamond or diamond nanoparticles are diamonds having particle size range less than one micrometre. Nano scale-size diamond fragments are propitious components for research due to their comparably narrow and small size allocation, flexible surface complex and chemical dormancy, altogether which constitute them as affirmative elements for various biological and electronic applications (Greiner *et al.*, 1988). It is foreseen that the properties of NDs will be utilized in an identical fashion to other quantum dots, carbon and metallic nano-particles for the creation of curative agents for transmission vehicles, probes, anti-bacterial and anti-viral treatments, gene therapy, tissue scaffolds, and medical devices like nano-robots (Freitas,



2003).

MATERIALS AND METHOD

Materials

Nanodiamond powder (ND) was procured from Sigma Aldrich (USA). Doxorubicin (DOX) was obtained as generous gift sample from Dabur India Limited, Ghaziabad, India. N-hydroxysuccinimide (NHS), dialysis membrane, 1-ethyl-3-(3-dimethyl laminopropyl) carbodiimide hydrochloride (EDC-HCl), N, N' Dicyclohexyl Carbodiimide (DCC), Pluronic F-68 were purchased from Himedia laboratories, Mumbai, India. Acetone, ethanol, concentrated HNO₃, sulphuric acid, hydrochloric acid, isopropyl alcohol and acetonitrile purchased from Merck Limited, Mumbai, India. Other chemicals which are consumed are of investigative chemical grade and used as brought.

Methodology

Synthesis of carboxylated ND (ND-COOH)

The commercially available synthesized nanodiamond powder having particle size of <10 nm, were carboxylated by following the standard procedure (Zheng *et al.*, 2009). Primarily the obtained nanodiamond powder was treated with the mixture of H₂SO₄ and HNO₃ (3:1 v/v) at room temperature for 48 h and then diluted with the addition of deionized water and centrifuged at 900 rpm for 30 min to separate out the ND particles. The obtained particles were rinsed by using deionized H₂O. Then obtained solution was again centrifuged at 900 rpm to separate out the ND particle, which was further heated at 90°C for 2 h in 0.1 M NaOH solution. Then ND sample was heated again with 0.1 M HCl for 2 h at 90°C, and washed by using deionized water to solution became weakly acidic. The obtained carboxylated-NDs were separated and dried under vacuum for further procedure (Zheng *et al.*, 2009).

Synthesis of activated nanodiamond

Firstly, 50 mg of ND powder was dispersed in 10 mL of distilled water (5 mg/mL) and then 250 mg of EDC-HCl was added with continuous stirring (Remi, Mumbai, India) for 12 h and maintain the pH 5.8 with the addition of 0.1N HCl. The EDC-HCl provides the effective activation of the carboxyl moiety of the NDs end group which enhances the attachment of NDs with another group of next chemical compound (-NH₂ moiety of the adipic acid dihydrazide). Subsequently addition of 125 mg of NHS into nanodiamond dispersion with constant rate of stirring and maintain the pH 5.8 with the addition of 0.1N HCl.

Synthesis of ND-DOX conjugate

To synthesize the different ND-DOX conjugate, the anticancer drugs (DOX) was conjugated with activated

carboxylated NDs (ND-COOH). Primarily, the anticancer drug (DOX) was dissolved in aqueous system (distilled water) according to their solubility parameters. DOX (50 mg) was dispersed in 10 mL of aqueous medium, then the prepared drug solution was activated with the addition of DCC and NHS. Surfactant (Pluronic F-68) medium of different concentration (0.25%, 0.5% and 1%) was formed by dissolving Pluronic-F-68 in acetone. 10 mL of drug dispersion (DOX) and ND-COOH (5 mg/mL) was added in drop wise manner in alternating sequence into separate Pluronic-F-68 (surfactant) medium with continuous stirring for 12 h to formulate drug conjugated nanodiamond i.e. ND-DOX conjugate. All the reaction mixtures were dialyzed to remove unreacted ND, DOX from all the formulated ND-DOX conjugate. The obtained different ND-DOX was separated by membrane filter (0.45 μm) and centrifuged at 15,000 rpm for 30 min (Remi, Mumbai, India). After centrifugation discarded the supernatant and lyophilized and preferred for subsequent analysis.

Characterization parameters of ND-DOX conjugate

The ND-DOX conjugate was prepared and authenticated by different spectroscopic instrumental methods i.e. nuclear magnetic resonance (¹H-NMR) (Bruker AvII-400), and fourier transform infrared (FTIR) (8400S, Shimadzu) spectroscopic technique. FTIR is most common technique used to determine the different functional group present in the prepared nanodiamond conjugate.

Particle morphology, particle size characteristic and surface charge (zeta potential)

The size of ND-DOX conjugate and the surface charge characteristics of synthesized ND-DOX conjugate were done by HORIBA SZ-100 series, (Horiba Scientific, Kyoto, Japan). The surface morphology of formulated ND-DOX conjugate was also be analyzed with Atomic Force Microscopic method (AFM) (Alpha300RA AFM, WITec, Germany). For taking AFM photomicrograph of the ND-DOX dispersion was spread on glass substrate on AC mode.

Drug loading efficiency of ND-DOX conjugate

The loading proficiency of different synthesized ND-DOX conjugate by HPLC system. The anticancer bioactive (DOX) conjugated nanodiamond were separately dispersed in 10 mL of PBS (pH 7.4) medium then the conjugate dispersed medium was filled in centrifuge tube and then centrifuged at 15000 rpm in cooling centrifuge (C-24BL, Remi, Mumbai, India) for 5 min. Then after centrifugation the supernatant were taken out and analyzed by using high performance liquid chromatography (HPLC) (Shimadzu, Japan). The HPLC system consist reverse phase Cosmosil



C18 column (4.6 mm × 250 mm, 5 μm, China) with UV detector was used to analyze the entrapment efficiency of prepared ND-DOX conjugate. The temperature of column should be maintained at 30 °C, for detection of DOX content, methanol: phosphate buffer (pH 7.4) at (70:30 v/v) was taken as mobile phase with the flow rate 0.8 mL/min and detection of DOX was done at 485 nm. The amount of drug loaded in ND was calculated from unbound drug in the supernatant, the loading efficiency of ND-DOX conjugate was analysed.

In-vitro release profile

The *in-vitro* drug release characteristics of the prepared ND-DOX conjugate were performed by modified dissolution technique. The prepared conjugate were placed separately in dialysis bag (Himedia Lab Limited, Mumbai, India) and dipped into the separate container containing 10 mL of phosphate buffer saline (PBS) having different pH range (pH 7.4, 6.5 and 5.5) and the medium was stirred with the rate of 100 rpm at 37±0.5°C. At a definite time interval an adequate quantity of the medium was taken out and replaced with the same quantity of the fresh PBS medium to maintain the sink condition. The collected medium was centrifuged at 5000 rpm for 15 min, subsequently the supernatant was taken out from the centrifuge tube and analyzed by HPLC system as previously described to carry out the amount of percentage cumulative drug released from ND-DOX conjugate (Dinarvand *et al.*, 2008).

Hemolytic toxicity

For hemolytic activity, whole human blood was amassed and collected in a collection vial from authorized pathology centre as denoted in Bhadra *et al.*, 2005. Firstly human blood was centrifuged at 10,000 rpm for 15 min for complete separation of RBC and plasma. The plasma was discarded and RBC was taken for further procedure. The red blood corpuscles (RBCs) (1 mL) was incubated separately with 10 mL of phosphate buffer solution pH 7.4 (taken as 100% hemolytic standard). In this hematocrit solution the ND-DOX conjugate and unmodified anticancer drugs (DOX), were added separately on hematocrit solution. The collection tube was allowed to stand for 1-2 h at 37°C, after that the ND-DOX conjugate in hematocrit mixture was centrifuged at 5000 rpm for 10 min, and then the absorbance was taken of supernatant at 540 nm to optimize the effect of ND-DOX conjugate and plain (unmodified) drugs against RBCs, which was useful to predict the percentage hemolysis.

SRB (sulforhodamine B) assay

Initially the cell line were grown in the RPMI 1640 medium containing fetal bovine serum (FBS 10%) and L-glutamine (2mM) further cell were inoculated on 96 well microtiter well plate containing 100 μL plating

densities. After it microtiter plates were incubated at temperature 37°C, 5% CO₂, 100% relative humidity and 95% air for 24 h before addition of formulations (i.e. ND-DOX conjugate, ND, plain anticancer drugs). Firstly conjugate were dissolved in dimethyl sulfoxide at 100 mg/ml further diluted to 1 mg/ml using distilled and kept frozen before use. Aliquots for frozen concentrate diluted to 100 μg/ml, 200 μg/ml, 400 μg/ml and 800 μg/ml during the addition of different conjugate with total medium consisting test articles. Final concentration of 10 μg/ml, 20 μg/ml, 40 μg/ml, 80 μg/ml was prepared by adding 10 μl of different dilutions in microtiter wells consisting 90 μl of medium.

Plates were incubated after compound was added at standard conditions for 48 h and assay was performed with cold trichloroacetic acid (TCA) and incubated at 4°C for 60 minutes further the supernatant was discarded and plates were cleaned. After than all the wells were treated with Sulforhodamine B (SRB) solution (50 μl) at 0.4 % (w/v) in 1 % acetic acid and incubated at room temperature for 30 minutes. The plates were dried and the absorbance was noted at 540 nm with 690 nm reference wavelength. Percent growth of cell was determined against control wells by plate-by-plate basis and further calculated as: $[Ti/C] \times 100 \%$. Finally the morphology of MCF-7 celllines was determined by the light microscope (Vichai *et al.*, 2006; Skehn *et al.*, 1990).

Cell Cycle Assay

MCF7 cells were developed on the 24-well plates with indicated ND-DOX conjugate for 48 hrs. After 48 hrs, adherent and non-adherent cells were gathered and settled with 70% cold ethanol at 4°C overnight. Later the cells were trypsinized with one microgram/ml of Hoechst for 20 minutes. The cells were rinsed and investigated by employing FACS. Then the cells were stimulated with 405m laser line and emission at 430/30 was collected for the DNA cell cycle study (Yuan *et al.*, 2016).

Statistical analysis

Final research outcome were showed as mean ± SD. All processes were performed thrice.

RESULTS AND DISCUSSION

The main objective behind the current study was to synthesize different ND-DOX conjugate for improved the targeting toward tumor cell as well as enhancing the potency of anticancer drug to provide maximum therapeutic effect. The surface of nanodiamond can be modified by many functional groups for providing higher stability of conjugate.

Spectroscopic analysis (¹H-NMR, and FTIR)



In $^1\text{H-NMR}$ spectrum of ND-DOX conjugate different distinctive peak was obtained. The proton assignment of ND was obtained at 1.02 ppm (2H, S) and 1.04 ppm (2H, S), shows presence of alkyl group, the peak at 2.17 (1H, S) shows hydroxyl moiety (R-OH). The proton assignment of DOX was obtained at ppm 1.2 ppm (3H, S) and 3.5 ppm (3H, S), 5.9 (1H, S) and 7.8 (1H, S). Beside of these proton assignment there are additional major peak was observed at 8.4 ppm (1H, S) and 8.9 ppm (1H, S), justify the presence of amide, which formed due to carbodiimide conjugation of carboxyl moiety of ND and amine group of DOX. All the proton assignment justifies the presence ND and DOX in the conjugate.

The FTIR spectra of ND-DOX conjugate were shown distinctive peak at 727 cm^{-1} , 857 cm^{-1} , 927 cm^{-1} , 2919 cm^{-1} shows C-H stretch, at 1285 cm^{-1} attribute to presence of C-O-C stretching, at 1588 cm^{-1} has a predominantly mixed C=C stretch and C=O stretch, at 1650 cm^{-1} displayed the presence of C=O stretch, a broad spectrum at 3250 cm^{-1} shows the presence of N-H stretch.

Surface characteristics

The surface characteristics like shape, size and texture of ND-DOX conjugate was examined by using atomic force microscopic technique (Alpha300RA AFM, WITec, Germany). The photomicrograph shows the uniform arrangement as well as similar height of the different ND-DOX conjugate. Summarizing, we have

shown that the prepared ND-DOX conjugate displayed flake-like structure with preferred crystal orientation as well as its also clearly depicted that the particles are certainly not all the same. The photomicrograph from AFM, it was displayed that some particles show clearer sharp edges than others, according to these finding we assume, however, that sharp edges on that length scale would be quite reactive.

Zeta potential determination, particle size analysis and drug entrapment proficiency

According to the result depicted in Table 1, the particle size evaluation of ND-DOX conjugate was carried out using particle size analyzer and the particle size of ND-DOX conjugate were observed 59.2 nm. All the data obtained from particle size analysis, it was confirm that the ND-DOX conjugate having nanometric size range. Zeta potential of ND-DOX conjugate was obtained to be -14.7 mV. The negative value of the zeta potential analysis of ND conjugate may be due to the carboxyl moiety of the nanodiamond. In the charge particle having increased zeta potential cause formation of more stable particles due to higher repulsive interaction. It was observed that the lower negative zeta potential may increase the stability of the system (Honary *et al.*, 2013). The loading efficiency of unconjugated (unbound) anticancer drug measured by HPLC and found to $94.3 \pm 1.25\%$.

Table 1 Ingredients and concentration using in the formulation of ND-DOX conjugate.

Formulation	Nanodiamond Concentration (mg)	Distilled water (ml)	DOX	Acetone/Ethanol (ml)	Particle size (nm)	Zeta potential (mV)	% Drug loading Efficiency
ND-DOX Conjugate	50	10	50	10	59.2nm	-14.7mV	94.3±1.25

In-vitro drug release pattern

It is recognized that well-organized release of drug from a drug delivery system is essential for therapeutic action of most of the anticancer drug conjugated formulations. Assimilation of acidic environment between the carrier and drug capable the liberation of an anticancer drug from the carrier (ND) into the cancer cell (having slightly acidic environment (pH 6.5)) then after endocytosis phase in the endosomes consisting pH range 5-6 and lysosomes (pH 4-5) of cancer cells. For this circumstances, the percentage release of drug from ND-DOX conjugate was performed at 37°C under simulated physiological conditions (phosphate-buffered saline, pH 7.4) and an acidic environment (phosphate-buffered saline, pH 5.5, an acidic endosome environment and pH 6.5, a simulated tumor environment) to analyse the feasibility of ND-DOX conjugate as an drug delivery system for anticancer drug. The amount and rate of drug released from ND-DOX conjugate were dependent on the pH of

the medium. ND-DOX conjugate displayed a rapid release pattern of anticancer drugs at pH 5.5 then pH 6.5 and then at pH 7.4. The release rate of drug from different ND-DOX conjugate after 96 hrs was found to be 35.5% at pH 5.5, whereas 24.6% was found at pH 6.5 and 16.2% of drug release was obtained at pH 7.4 correspondingly. Drug conjugated with ND via amide bond which is relatively more stable than physical absorption, thus slowing down the release of anticancer drug from ND conjugate (Zhu *et al.*, 2017; Hu *et al.*, 2010). This is due to because of surface modification of ND via carbodiimide conjugation between of the carboxyl group of nanodiamond with the amine group of adipic acid dihydrazide and formation of crosslinked core-shell micelle, which having less solubility, may promote sustained release (Ansari *et al.*, 2016). The data also has been suggested by various researchers, that acidic pH condition triggered the cleavage of amide bond (Zhao *et al.*, 2014). The slow rate of anticancer drug (DOX) release from ND-DOX conjugate was



calculated at pH 7.4, which mimics the physiological environment of the bloodstream and establish that reduced amount of anticancer drugs is liberated from conjugate in the blood circulation. It was assume that ND-DOX conjugate would adhere preferentially in the site of tumor cell via enhanced permeability and retention (EPR) effect. The unmodified anticancer drug (DOX), and the plateaus by approximately 30 mins at pH 7.4, 6.5 and 5.5 respectively, with a highest release of 98.5% at pH 7.4, the release 96.7% occurs at pH 6.5, and 94.8% at pH 5.5 correspondingly. In contrast with ND-DOX conjugate, the free or unmodified anticancer drugs cause damage cells in a normal physiological environment, causing serious side effects.

Hemolytic toxicity study

The hemotoxic effect of the prepared ND-DOX conjugate was estimated by hemolytic toxicity study. The ND-DOX conjugate exhibited hemolytic toxicity upto, $5.18 \pm 0.25\%$. Whereas the plain drug (DOX) represents hemolytic toxicity $55.12 \pm 0.5\%$. There was decline in hemolytic toxicity by different ND-DOX conjugate in compare to plain drug, caused due to delayed release of encapsulated drug molecules in the nanodiamond conjugate. The repression of hemotoxicity of drug can be linked among other similar studies described previously (Wąsowicz *et al.*, 2017).

SRB (sulforhodamine B) assay

The *in-vitro* cytotoxicity screening of ND-DOX conjugate in human breast cancer cellline (MCF-7) was established by SRB assay. The result obtained by the assay affirm dose dependent assessment of cytotoxicity in which the cellular bioavailability decreased with increasing the concentration of sample (ND-DOX conjugate and anticancer drugs DOX). Which revealed that higher concentration of sample ND-DOX conjugate inhibit the cell growth. Furthermore, the cell viability also gets declined with increase in the concentration of sample. ND-DOX conjugate formulation were experiential to be cytotoxic to a greater amount with the concentration between 10-80 $\mu\text{g/ml}$.

Cell Cycle Assay

The antitumor mechanisms of prepared ND-DOX conjugates have consorted with mitosis, apoptosis and cell cycle arrest in the G2/M phase (Zhang *et al.*, 2010). Consequently, the enhancement in G2/M phase arrest, signify the improved beneficial and therapeutic effect of conjugate. The effect of the arrest of G2/M phase (cell cycle) was investigated by the various range of concentration of conjugates against MCF-7 cells (i.e. 20 $\mu\text{g/ml}$, 40 $\mu\text{g/ml}$ and 80 $\mu\text{g/ml}$). These conclusions suggested that ND-DOX conjugate at different concentration could encourage cell cycle arrest in MCF-7 cells.

Table 2. Cell cycle analysis: Percentage of cells in different cell cycle stages vs MCF-7.

S. No.	Cell Cycle stage	Cell Control	ND-DOX
1	Sub G0/G1	0.6	2.45
2	G0/G1	74.49	59.15
3	S	6.11	9.7
4	G2/M	18.94	27.26

Note: Table showing the % of cells get arrested in the different stages of their life cycle. In Sub G0/G1 phase (Apoptotic phase), 0.6% and 2.45% of cells get arrested in Untreated and test compound ND-DOX respectively. In G0/G1 phase (Growth Phase), 74.49% and 59.15% of cells get arrested in Untreated and test compound ND-DOX respectively. In S phase (synthetic phase), 6.11% and 9.7% of cells get arrested in Untreated and test compound ND-DOX respectively. On the other hand, in G2/M phase, 18.94% and 27.26% of cells get arrested in Untreated and test compound ND- DOX respectively.

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